

# PIPEGRIP TYPE N BLUE SOLVENT CEMENT

### **RLA Polymers Pty Ltd**

Chemwatch: **5226-45** Version No: **3.1.1.1** 

Safety Data Sheet according to WHS and ADG requirements

#### Chemwatch Hazard Alert Code: 3

Issue Date: **04/10/2016** Print Date: **09/12/2016** S.GHS.AUS.EN

### SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

### **Product Identifier**

Product name	PIPEGRIP TYPE N BLUE SOLVENT CEMENT	
Synonyms	Not Available	
Proper shipping name	DHESIVES containing flammable liquid	
Other means of identification	Not Available	

## Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses

The use of a quantity of material in an unventilated or confined space may result in increased exposure and an irritating atmosphere developing. Before starting consider control of exposure by mechanical ventilation.

Use according to manufacturer's directions.

Type N Solvent Cement for non-pressure joints in PVC-U Pipes and Fittings.

### Details of the supplier of the safety data sheet

Registered company name	RLA Polymers Pty Ltd	
Address	15 Colchester Road Kilsyth VIC 3137 Australia	
Telephone	+61 3 9728 1644	
Fax	+61 3 9728 6009	
Website	www.rlagroup.com.au	
Email	sales@rlagroup.com.au	

## Emergency telephone number

	Association / Organisation		
	Emergency telephone numbers		
	Other emergency telephone numbers	132766 (Security Monitoring Service)	

## **SECTION 2 HAZARDS IDENTIFICATION**

### Classification of the substance or mixture

Poisons Schedule	Not Applicable  Flammable Liquid Category 2, Eye Irritation Category 2A, Reproductive Toxicity Category 2, Specific target organ toxicity - single exposure Category 3 (respiratory tract irritation), Specific target organ toxicity - single exposure Category 3 (narcotic effects), Aspiration Hazard Category 1	
Classification <sup>[1]</sup>		
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HSIS ; 3. Classification drawn from EC Directive 1272/2008 - Annex VI	

## Label elements

GHS label elements



SIGNAL WORD	DANGER	
Hazard statement(s)		
H225	Highly flammable liquid and vapour.	
H319 Causes serious eye irritation.		
H361	Suspected of damaging fertility or the unborn child.	
H335	May cause respiratory irritation.	
H336	May cause drowsiness or dizziness.	

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May be fatal if swallowed and enters airways.	
May form explosive peroxides	
Repeated exposure may cause skin dryness and cracking	

## Precautionary statement(s) Prevention

P201	Obtain special instructions before use.	
P210	Keep away from heat/sparks/open flames/hot surfaces No smoking.	
P271	Use only outdoors or in a well-ventilated area.	
P281	Use personal protective equipment as required.	

## Precautionary statement(s) Response

P301+P310	IF SWALLOWED: Immediately call a POISON CENTER or doctor/physician.	
P308+P313	308+P313 IF exposed or concerned: Get medical advice/attention.	
P331	Do NOT induce vomiting.	
P370+P378	In case of fire: Use alcohol resistant foam or normal protein foam for extinction.	

### Precautionary statement(s) Storage

P403+P235	Store in a well-ventilated place. Keep cool.
P405	Store locked up.

## Precautionary statement(s) Disposal

P501 Dispose of contents/container in accordance with local regulations.

### **SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS**

### Substances

See section below for composition of Mixtures

#### Mixtures

CAS No	%[weight]	Name
78-93-3	>60	methyl ethyl ketone
108-94-1	10-30	cyclohexanone
109-99-9	<5	<u>tetrahydrofuran</u>
	balance	Ingredients determined not to be hazardous

## **SECTION 4 FIRST AID MEASURES**

## Description of first aid measures

Eye Contact	If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	If fumes or combustion products are inhaled remove from contaminated area.  Lay patient down. Keep warm and rested.  Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.  Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.  Transport to hospital, or doctor, without delay.
Ingestion	If swallowed do NOT induce vomiting.  If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.  Observe the patient carefully.  Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.  Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.  Seek medical advice.  If spontaneous vomiting appears imminent or occurs, hold patient's head down, lower than their hips to help avoid possible aspiration of vomitus.

## Indication of any immediate medical attention and special treatment needed

Any material aspirated during vomiting may produce lung injury. Therefore emesis should not be induced mechanically or pharmacologically. Mechanical means should be used if it is considered necessary to evacuate the stomach contents; these include gastric lavage after endotracheal intubation. If spontaneous vomiting has occurred after ingestion, the patient should be monitored for difficult breathing, as adverse effects of aspiration into the lungs may be delayed up to 48 hours.

Treat symptomatically	۲.
for simple ketones:	

BASIC TREATMENT

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Establish a patent airway with suction where necessary.

- Watch for signs of respiratory insufficiency and assist ventilation as necessary
- Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- Monitor and treat, where necessary, for pulmonary oedema.
- Monitor and treat, where necessary, for shock
- DO NOT use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5mL/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.
- Give activated charcoal

## ADVANCED TREATMENT

- ▶ Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred
- Consider intubation at first sign of upper airway obstruction resulting from oedema.
- ▶ Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- ▶ Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema.
- ▶ Hypotension with signs of hypovolaemia requires the cautious administration of fluids. Fluid overload might create complications.
- Treat seizures with diazepam.
- Proparacaine hydrochloride should be used to assist eye irrigation.

#### **EMERGENCY DEPARTMENT**

- Laboratory analysis of complete blood count, serum electrolytes, BUN, creatinine, glucose, urinalysis, baseline for serum aminotransferases (ALT and AST), calcium, phosphorus and magnesium, may assist in establishing a treatment regime. Other useful analyses include anion and osmolar gaps, arterial blood gases (ABGs), chest radiographs and electrocardiograph.
- Positive end-expiratory pressure (PEEP)-assisted ventilation may be required for acute parenchymal injury or adult respiratory distress syndrome.
- ▶ Consult a toxicologist as necessary

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

### **SECTION 5 FIREFIGHTING MEASURES**

### **Extinguishing media**

- Alcohol stable foam.
- Dry chemical powder
- BCF (where regulations permit)
- Carbon dioxide.

### Special hazards arising from the substrate or mixture

Fire Incompatibility	▶ Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
Advice for firefighters	
Fire Fighting	Alert Fire Brigade and tell them location and nature of hazard. May be violently or explosively reactive. Wear breathing apparatus plus protective gloves in the event of a fire. Prevent, by any means available, spillage from entering drains or water course.
Fire/Explosion Hazard	Liquid and vapour are highly flammable.  Severe fire hazard when exposed to heat, flame and/or oxidisers.  Vapour may travel a considerable distance to source of ignition.  Heating may cause expansion or decomposition leading to violent rupture of containers.  Combustion products include: carbon dioxide (CO2) other pyrolysis products typical of burning organic material.  Contains low boiling substance: Closed containers may rupture due to pressure buildup under fire conditions.  WARNING: Long standing in contact with air and light may result in the formation

## **SECTION 6 ACCIDENTAL RELEASE MEASURES**

HAZCHEM

## Personal precautions, protective equipment and emergency procedures

of potentially explosive peroxides

See section 8

## **Environmental precautions**

See section 12

## Methods and material for containment and cleaning up

Minor Spills	Remove all ignition sources. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment.
Major Spills	Clear area of personnel and move upwind.  Alert Fire Brigade and tell them location and nature of hazard.  May be violently or explosively reactive.  Wear breathing apparatus plus protective gloves.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

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### **SECTION 7 HANDLING AND STORAGE**

#### Precautions for safe handling

Containers, even those that have been emptied, may contain explosive vapours.

Do NOT cut, drill, grind, weld or perform similar operations on or near containers.

May form explosive peroxides on standing or following concentration by distillation.

Review of stocks and testing for peroxide content by given tested procedures at 3-monthly intervals is recommended, together with safe disposal of peroxidic

[Peroxide-containing residues can often be rendered innocuous by pouring into an excess of sodium carbonate solution]

Contains low boiling substance:

Storage in sealed containers may result in pressure buildup causing violent rupture of containers not rated appropriately. Safe handling

Check for bulging containers.

Vent periodically

Always release caps or seals slowly to ensure slow dissipation of vapours

DO NOT allow clothing wet with material to stay in contact with skin

Avoid all personal contact, including inhalation.

Wear protective clothing when risk of exposure occurs

Use in a well-ventilated area.

Prevent concentration in hollows and sumps.

Other information

Store in original containers in approved flame-proof area.

No smoking, naked lights, heat or ignition sources. DO NOT store in pits, depressions, basements or areas where vapours may be trapped.

Keep containers securely sealed.

## Conditions for safe storage, including any incompatibilities

Suitable container	Glass container is suitable for laboratory quantities Packing as supplied by manufacturer. Plastic containers may only be used if approved for flammable liquid. Check that containers are clearly labelled and free from leaks. For low viscosity materials (i): Drums and jerry cans must be of the non-removable head type. (ii): Where a can is to be used as an inner package, the can must have a screwed enclosure. For materials with a viscosity of at least 2680 cSt. (23 deg. C) For manufactured product having a viscosity of at least 250 cSt.
	Avoid strong bases.  The unhindered oxygen atom found on cyclic ethers such as the epoxides, oxetanes, furans, dioxanes and pyrans, carries two unshared pairs of electrons - a structure which favors the formation of coordination complexes and the solvation of cations.

Storage incompatibility

Cyclic ethers are used as important solvents, as chemical intermediate and as monomers for ring-opening polymerization.

They are unstable at room temperature due to possibility of peroxide formation; stabiliser is sometimes needed for storage and transportation.

NOTE: Ethers lacking non-methyl hydrogen atoms adjacent to the ether link are thought to be relatively safe

Avoid reaction with oxidising agents

## **SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION**

## **Control parameters**

## OCCUPATIONAL EXPOSURE LIMITS (OEL)

## INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	methyl ethyl ketone	Methyl ethyl ketone (MEK)	445 mg/m3 / 150 ppm	890 mg/m3 / 300 ppm	Not Available	Not Available
Australia Exposure Standards	cyclohexanone	Cyclohexanone	100 mg/m3 / 25 ppm	Not Available	Not Available	Sk
Australia Exposure Standards	tetrahydrofuran	Tetrahydrofuran	295 mg/m3 / 100 ppm	Not Available	Not Available	Sk

### **EMERGENCY LIMITS**

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
methyl ethyl ketone	Butanone, 2-; (Methyl ethyl ketone; MEK)	Not Available	Not Available	Not Available
cyclohexanone	Cyclohexanone; (Ketohexamethylene)	60 ppm	830 ppm	5000 ppm
tetrahydrofuran	Tetrahydrofuran	Not Available	Not Available	Not Available

Ingredient	Original IDLH	Revised IDLH
methyl ethyl ketone	3,000 ppm	3,000 [Unch] ppm
cyclohexanone	5,000 ppm	700 ppm
tetrahydrofuran	20,000 [LEL] ppm	2,000 [LEL] ppm

### **Exposure controls**

#### Appropriate engineering controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and

"removes" air in the work environment.

### Personal protection Safety glasses with side shields Chemical goggles. Eye and face protection Contact lenses may pose a special hazard: soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. Skin protection See Hand protection below Wear chemical protective gloves, e.g. PVC. Wear safety footwear or safety gumboots, e.g. Rubber The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior Hands/feet protection to the application. The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice. Personal hygiene is a key element of effective hand care. **Body protection** See Other protection below Overalls. PVC Apron PVC protective suit may be required if exposure severe. Eyewash unit. Other protection · Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity. · For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets). Non sparking safety or conductive footwear should be considered. Conductive footwear describes a boot or shoe with a sole made from a conductive compound chemically bound to the bottom components, for permanent control to electrically ground the foot an shall dissipate static electricity from the body to reduce the possibility of ignition of volatile compounds.

#### Recommended material(s)

#### GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the

"Forsberg Clothing Performance Index".

Thermal hazards

The effect(s) of the following substance(s) are taken into account in the computergenerated selection:

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Material	СРІ
PE/EVAL/PE	A
BUTYL	С
BUTYL/NEOPRENE	С
CPE	С
HYPALON	С
NATURAL RUBBER	С
NATURAL+NEOPRENE	С
NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
NITRILE+PVC	С
PVA	С
PVC	С
SARANEX-23	С
TEFLON	С
VITON/CHLOROBUTYL	С
VITON/NEOPRENE	С

<sup>\*</sup> CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

#### Respiratory protection

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 5 x ES	A-AUS / Class 1	-	A-PAPR-AUS / Class 1
up to 25 x ES	Air-line*	A-2	A-PAPR-2
up to 50 x ES	-	A-3	-
50+ x ES	-	Air-line**	-

### ^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content. The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.

## **SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES**

### Information on basic physical and chemical properties

**Appearance** Blue viscous highly flammable liquid with a characteristic odour of MEK; does not mix with water.

Not Available

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Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	515
pH (as supplied)	Not Applicable	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	-16	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	HIGHLY FLAMMABLE.	Oxidising properties	Not Available
Upper Explosive Limit (%)	10	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	1.8	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Immiscible	pH as a solution (1%)	Not Applicable
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

# **SECTION 10 STABILITY AND REACTIVITY**

Reactivity	See section 7
Chemical stability	Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

# **SECTION 11 TOXICOLOGICAL INFORMATION**

# Information on toxicological effects

Inhaled	The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.  Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by sleepiness, reduced alertness, loss of reflexes, lack of co-ordination, and vertigo.  Inhalation hazard is increased at higher temperatures.  Inhalation of high concentrations of gas/vapour causes lung irritation with coughing and nausea, central nervous depression with headache and dizziness, slowing of reflexes, fatigue and inco-ordination.  Inhalation of aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual.
Ingestion	Swallowing of the liquid may cause aspiration into the lungs with the risk of chemical pneumonitis; serious consequences may result. (ICSC13733) Ingestion of tetrahydrofuran may not, in itself, produce internal injury, however, contaminating levels of furan, present in certain grades of commercial product, may produce liver and kidney injury exacerbated by the intake of alcoholic beverages.  Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.
Skin Contact	Repeated exposure may cause skin cracking, flaking or drying following normal handling and use.  There is some evidence to suggest that the material may cause moderate inflammation of the skin either following direct contact or after a delay of some time.  Repeated exposure can cause contact dermatitis which is characterised by redness, swelling and blistering.  Open cuts, abraded or irritated skin should not be exposed to this material  Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.  Skin contact with the material may be harmful; systemic effects may result following absorption.
Еуе	The vapour when concentrated has pronounced eye irritation effects and this gives some warning of high vapour concentrations. If eye irritation occurs seek to reduce exposure with available control measures, or evacuate area.  There is evidence that material may produce eye irritation in some persons and produce eye damage 24 hours or more after instillation. Severe inflammation may be expected with pain.
Chronic	Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems. Based on experience with animal studies, exposure to the material may result in toxic effects to the development of the foetus, at levels which do not cause significant toxic effects to the mother.  Prolonged or repeated skin contact may cause drying with cracking, irritation and possible dermatitis following.  There has been some concern that this material can cause cancer or mutations but there is not enough data to make an assessment.  Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.  Long term cyclohexanone exposure may cause liver and kidney changes. Clouding of the eye lens and cataract development may occur.  Limited information is available on the chronic (long-term) effects of methyl ethyl ketone in humans. Chronic inhalation studies in animals have reported slight neurological, liver, kidney, and respiratory effects. No information is available on the developmental, reproductive, or carcinogenic effects of methyl ethyl ketone in humans. Developmental effects, including decreased foetal weight and foetal malformations, have been reported in mice and rats exposed to methyl ethyl ketone via inhalation and ingestion.

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TYPE N BLUE SOLVENT	TOXICITY	IRRITATION
CEMENT	Not Available	Not Available
	TOXICITY	IRRITATION
	Dermal (rabbit) LD50: >8100 mg/kg <sup>[1]</sup>	Eye (human): 350 ppm -irritant
methyl ethyl ketone	Inhalation (rat) LC50: 23.5 mg/L/8hr <sup>[2]</sup>	Eye (rabbit): 80 mg - irritant
	Inhalation (rat) LC50: 50.1 mg/L/8 hr <sup>[2]</sup>	Skin (rabbit): 402 mg/24 hr - mild
	Oral (rat) LD50: 3474.9 mg/kg <sup>[1]</sup>	Skin (rabbit):13.78mg/24 hr open
	тохісіту	IRRITATION
	Dermal (rabbit) LD50: 947.8 mg/kg <sup>[2]</sup>	Eye (human): 75 ppm
cyclohexanone	Inhalation (rat) LC50: 8000 ppm/4hr <sup>[2]</sup>	Eye (rabbit): 0.25 mg/24h SEVERE
	Oral (rat) LD50: 1535 mg/kg <sup>[2]</sup>	Eye (rabbit): 4.74 mg SEVERE
		Skin (rabbit): 500 mg(open) mild
	тохісіту	IRRITATION
	dermal (rat) LD50: >2000 mg/kg <sup>[1]</sup>	Not Available
	Inhalation (rat) LC50: >14.7 mg/l6 hr <sup>[1]</sup>	
tetrahydrofuran	Inhalation (rat) LC50: 2100 ppm/3hr <sup>[2]</sup>	
	Inhalation (rat) LC50: 21000 ppm/3hr <sup>[2]</sup>	
	Inhalation (rat) LC50: 72 mg/L/2hr <sup>[2]</sup>	
	Oral (rat) LD50: <891 mg/kg> <sup>[1]</sup>	
Legend:	Nature obtained from Europe ECHA Registered Substances - Acute toxicity 2.     extracted from RTECS - Register of Toxic Effect of chemical Substances	* Value obtained from manufacturer's SDS. Unless otherwise specified data
-		ethyl ketone is often used in combination with other solvents and the toxic effe
Legend:	extracted from RTECS - Register of Toxic Effect of chemical Substances  Methyl ethyl ketone is considered to have a low order of toxicity; however methyl	ethyl ketone is often used in combination with other solvents and the toxic effe h methyl ethyl ketone and also methyl n-butyl ketone with methyl ethyl ketone
-	extracted from RTECS - Register of Toxic Effect of chemical Substances  Methyl ethyl ketone is considered to have a low order of toxicity; however methyl of the mix may be greater than either solvent alone. Combinations of n-hexane wit show increase in peripheral neuropathy, a progressive disorder of nerves of extre	ethyl ketone is often used in combination with other solvents and the toxic effeth methyl ethyl ketone and also methyl n-butyl ketone with methyl ethyl ketone mities.  In this shave been noted at higher doses. Other features of toxicity include
METHYL ETHYL KETONE	extracted from RTECS - Register of Toxic Effect of chemical Substances  Methyl ethyl ketone is considered to have a low order of toxicity; however methyl of the mix may be greater than either solvent alone. Combinations of n-hexane wit show increase in peripheral neuropathy, a progressive disorder of nerves of extre Combinations with chloroform also show increase in toxicity  Cyclohexanone irritates the eye and the skin. Signs of CNS depression and weigi mottling of the lungs and degenerative changes in the liver and kidney. It is not of the substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.	ethyl ketone is often used in combination with other solvents and the toxic effeth methyl ethyl ketone and also methyl n-butyl ketone with methyl ethyl ketone imities.  In loss have been noted at higher doses. Other features of toxicity include considered to cause cancers, but it may reversibly reduce fertility.  The and may produce on contact skin redness, swelling, the production of
METHYL ETHYL KETONE  CYCLOHEXANONE	Methyl ethyl ketone is considered to have a low order of toxicity; however methyl of the mix may be greater than either solvent alone. Combinations of n-hexane wit show increase in peripheral neuropathy, a progressive disorder of nerves of extre Combinations with chloroform also show increase in toxicity  Cyclohexanone irritates the eye and the skin. Signs of CNS depression and weigi mottling of the lungs and degenerative changes in the liver and kidney. It is not of the substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Evidence of carcinogenicity may be inadequate or limited in animal testing.  The material may cause severe skin irritation after prolonged or repeated exposures may produce so	ethyl ketone is often used in combination with other solvents and the toxic effeth methyl ethyl ketone and also methyl n-butyl ketone with methyl ethyl ketone imities.  In loss have been noted at higher doses. Other features of toxicity include considered to cause cancers, but it may reversibly reduce fertility.  In and may produce on contact skin redness, swelling, the production of evere ulceration.  In the material ceases. This may be due to a non-allergenic condition known as sure to high levels of highly irritating compound. Key criteria for the diagnosis ndividual, with abrupt onset of persistent asthma-like symptoms within minutents in spirometry, with the presence of moderate to severe bronchial hyperreactivity.
CYCLOHEXANONE  TETRAHYDROFURAN  METHYL ETHYL KETONE & TETRAHYDROFURAN	Methyl ethyl ketone is considered to have a low order of toxicity; however methyl of the mix may be greater than either solvent alone. Combinations of n-hexane wit show increase in peripheral neuropathy, a progressive disorder of nerves of extre Combinations with chloroform also show increase in toxicity  Cyclohexanone irritates the eye and the skin. Signs of CNS depression and weigi mottling of the lungs and degenerative changes in the liver and kidney. It is not of the substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Evidence of carcinogenicity may be inadequate or limited in animal testing.  The material may cause severe skin irritation after prolonged or repeated exposur vesicles, scaling and thickening of the skin. Repeated exposures may produce so Oral (human) LDLo: 50 mg/kg* [CCINFO]* Nil reported  Asthma-like symptoms may continue for months or even years after exposure to reactive airways dysfunction syndrome (RADS) which can occur following expos of RADS include the absence of preceding respiratory disease, in a non-atopic to hours of a documented exposure to the irritant. A reversible airflow pattern, o on methacholine challenge testing and the lack of minimal lymphocytic inflamma	ethyl ketone is often used in combination with other solvents and the toxic effeth methyl ethyl ketone and also methyl n-butyl ketone with methyl ethyl ketone mities.  In loss have been noted at higher doses. Other features of toxicity include considered to cause cancers, but it may reversibly reduce fertility.  The and may produce on contact skin redness, swelling, the production of evere ulceration.  In the material ceases. This may be due to a non-allergenic condition known as ure to high levels of highly irritating compound. Key criteria for the diagnosis individual, with abrupt onset of persistent asthma-like symptoms within minute in spirometry, with the presence of moderate to severe bronchial hyperreactivity, ation, without eosinophilia, have also been included in the criteria for diagnosis
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O - Data Not Available to make classification

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### **SECTION 12 ECOLOGICAL INFORMATION**

#### Toxicity

Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
methyl ethyl ketone	LC50	96	Fish	228.130mg/L	3
methyl ethyl ketone	EC50	48	Crustacea	308mg/L	2
methyl ethyl ketone	EC50	96	Algae or other aquatic plants	>500mg/L	4
methyl ethyl ketone	EC50	384	Crustacea	52.575mg/L	3
methyl ethyl ketone	NOEC	48	Crustacea	68mg/L	2
cyclohexanone	LC50	96	Fish	71.940mg/L	3
cyclohexanone	EC50	72	Algae or other aquatic plants	32.9mg/L	5
cyclohexanone	EC10	72	Algae or other aquatic plants	3.56mg/L	4
cyclohexanone	NOEC	24	Fish	ca.5mg/L	1
tetrahydrofuran	LC50	96	Fish	72.742mg/L	3
tetrahydrofuran	EC50	96	Algae or other aquatic plants	310.515mg/L	3
tetrahydrofuran	EC50	384	Crustacea	17.029mg/L	3
tetrahydrofuran	NOEC	24	Fish	>=5mg/L	1
I edend:			A Registered Substances - Ecotoxicological Info	, ,	

Legend:

Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

### DO NOT discharge into sewer or waterways

### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
methyl ethyl ketone	LOW (Half-life = 14 days)	LOW (Half-life = 26.75 days)
cyclohexanone	LOW	LOW
tetrahydrofuran	LOW	LOW

### **Bioaccumulative potential**

Ingredient	Bioaccumulation	
methyl ethyl ketone	LOW (LogKOW = 0.29)	
cyclohexanone	LOW (BCF = 2.45)	
tetrahydrofuran	LOW (LogKOW = 0.46)	

## Mobility in soil

Ingredient	Mobility	
methyl ethyl ketone	MEDIUM (KOC = 3.827)	
cyclohexanone	LOW (KOC = 15.15)	
tetrahydrofuran	LOW (KOC = 4.881)	

## **SECTION 13 DISPOSAL CONSIDERATIONS**

### Waste treatment methods

Product / Packaging

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- Reduction
- Reuse
- Recycling
- Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use.

- ▶ DO NOT allow wash water from cleaning or process equipment to enter drains.
- ▶ It may be necessary to collect all wash water for treatment before disposal.
- ▶ In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- ▶ Where in doubt contact the responsible authority.
- Recycle wherever possible
- Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
- Dispose of by: burial in a land-fill specifically licenced to accept chemical and / or pharmaceutical wastes or Incineration in a licenced apparatus (after admixture with suitable combustible material).
- ▶ Decontaminate empty containers.

## **SECTION 14 TRANSPORT INFORMATION**

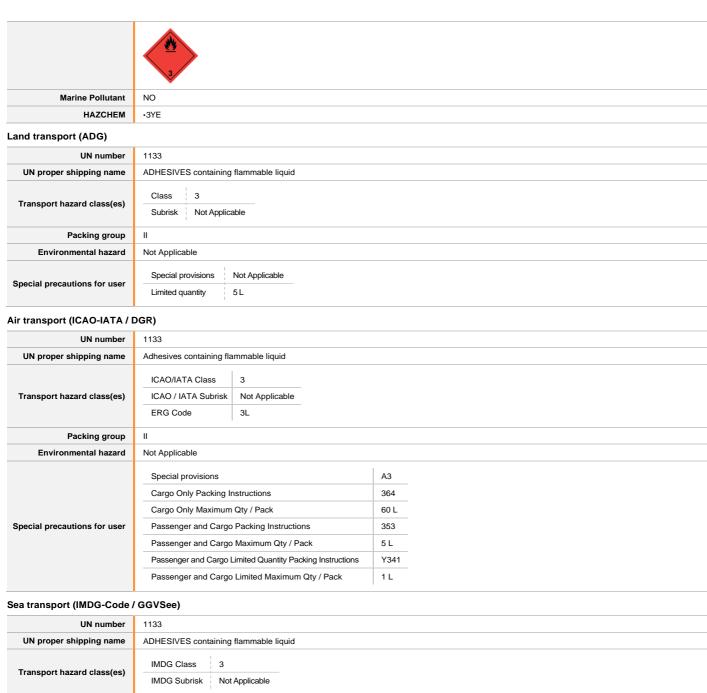
disposal

## **Labels Required**

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UN number	1133		
UN proper shipping name	ADHESIVES containing flammable liquid		
Transport hazard class(es)	IMDG Class 3 IMDG Subrisk Not Applicable		
Packing group	П		
Environmental hazard	Not Applicable		
Special precautions for user	EMS Number F-E, S-D Special provisions Not Applicable Limited Quantities 5 L		

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

# **SECTION 15 REGULATORY INFORMATION**

Safety, health and environmental regulations / legislation specific for the substance or mixture

 $\parallel$  METHYL ETHYL KETONE(78-93-3) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Exposure Standards

Australia Inventory of Chemical Substances (AICS)

Australia Hazardous Substances Information System - Consolidated Lists

CYCLOHEXANONE(108-94-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

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## TYPE N BLUE SOLVENT CEMENT

Australia Exposure Standards Australia Inventory of Chemical Substances (AICS) Australia Hazardous Substances Information System - Consolidated Lists International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs TETRAHYDROFURAN(109-99-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Inventory of Chemical Substances (AICS)

Australia Hazardous Substances Information System - Consolidated Lists

National Inventory	Status	
Australia - AICS	Y	
Canada - DSL	Y	
Canada - NDSL	N (cyclohexanone; tetrahydrofuran; methyl ethyl ketone)	
China - IECSC	Υ	
Europe - EINEC / ELINCS / NLP	Y	
Japan - ENCS	Υ	
Korea - KECI	Υ	
New Zealand - NZIoC	Υ	
Philippines - PICCS	Υ	
USA - TSCA	Υ	
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)	

## **SECTION 16 OTHER INFORMATION**

#### Other information

Australia Exposure Standards

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

### **Definitions and abbreviations**

PC-TWA: Permissible Concentration-Time Weighted Average PC-

STEL: Permissible Concentration-Short Term Exposure Limit IARC:

International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value

LOD: Limit Of Detection

OTV: Odour Threshold Value

BCF: BioConcentration Factors

BEI: Biological Exposure Index

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TEL (+61 3) 9572 4700.